Matter in external magnetic field



http://www.ru.nl/hfml/research/levitation/diamagnetic/

$$B = B_0 + B'$$



 B_0 – external magnetic field B' – induced magnetic field

relative magnetic permeability: $\mu_r = B/B_0$

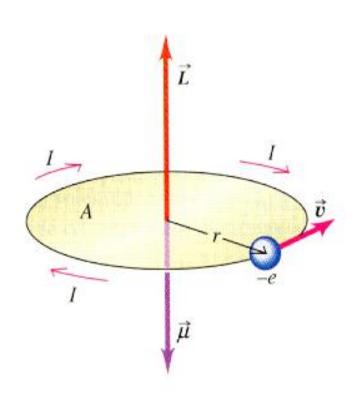
Matter in external magnetic field

relative magnetic permeability: $\mu_r = B/B_0$

$$\mu_r \ge 1$$
, $B \approx B_0 (1+10^{-4})$ paramagnetics $\mu_r \le 1$, $B \approx B_0 (1-10^{-6})$ diamagnetics $\mu_r >> 1$, $B \approx B_0 \cdot 10^3$ feromagnetics

Magnetic properties of atom

- Classical model of atom: electrons rotate around nucleus
- magnetic field is induced by "microcurrents"
- nucleus can has spin magnetic moment
- electron can has spin magnetic moment and orbital magnetic moment
- Classical model is not satisfactory



Angular moment: $\vec{L} = m \vec{r} \times \vec{v}$

Orbital magnetic moment:

$$\overrightarrow{\mu_l} = \frac{-e}{2m} \overrightarrow{L}$$

Spin magnetic moment:

$$\overrightarrow{\mu_S} = \frac{-e}{2m} \overrightarrow{S}$$

For simplicity: spin magnetic moment of nucleus is neglected.

Diamagnetism

- All types of matter have the property of diamagnetism.
- Dominant property of matter where molecules have paired electronic spins (no net spin) only magnetic moment from orbital motion is in interaction with the field.
- When external magnetic field is switched on induces change in orbital and magnetic moment of electron.
- Induced magnetic field is in opposite direction of external field (Lentz's rule)

Diamagnetism

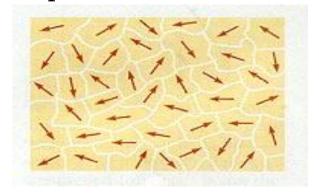
B' $\approx 10^{-6} \, \text{B}_0$

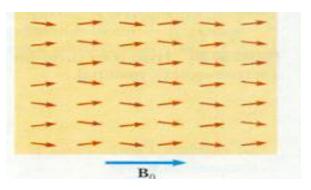
diamagnetics: noble gases, nitrogen, hydrogen, cooper, grafit, gold



Paramagnetism

- magnetic dipole moment: orbital and spin
- out of field spin magnetic moments are oriented randomly (chaotic)
- In the magnetic field they will tend to orientate in the direction parallel to field lines





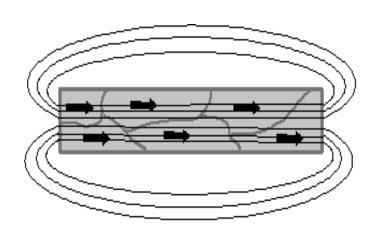
$$B' \approx 10^{-4} B_0$$

Paramagnetics: aluminium, manganese, sodium, potassium, free radicals

$$\mu_r \ge 1$$
, B \approx B₀ (1+10⁻⁴) paramagnetics

Feromagnetism





- Inside particular magnetic domain, spin magnetic moments are mutually parallel
- Orientation of different domains is not the same before first magnetization; when the magnetic field is applied, spin magnetic moments in amm domains tend to align along a field
- When B0 is switched of, ordering of domains (magnetization) remains

B'
$$\approx 10^3 \, \text{B}_0$$

Feromagnetics: metals VIII – X group



Figure 2 Levitation at your fingertips. A strong NdFeB magnet (14 tesla) levitates 2.5 metres below a powerful superconducting magnet. The field at the levitation point is about 500 Gauss.

$$B = B_0 + B'$$

$$\mu_r = B/B_0$$

$$\mu_r \ge 1$$
, B $\approx B_0 (1+10^{-4})$ paramagnetics

$$\mu_r \le 1$$
, B \approx B₀ (1-10⁻⁶) diamagnetics

$$\mu_{\rm r} >> 1$$
, B \approx B₀·10³

feromagnetics

spin magnetic moment

para and feromagnetism

magnetic moment of electron

and/only

orbital magnetic moment diamagnetism